

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
11 December 2003 (11.12.2003)

PCT

(10) International Publication Number  
WO 03/102044 A2

(51) International Patent Classification<sup>7</sup>: C08G 59/68,  
C07C 275/14, C08L 63/00, C08G 18/28, C09D 5/03

(21) International Application Number: PCT/US03/14366

(22) International Filing Date: 6 May 2003 (06.05.2003)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
10/160,466 31 May 2002 (31.05.2002) US

(71) Applicant: PPG INDUSTRIES OHIO INC. [US/US];  
3800 West 143rd Street, Cleveland, OH 44111 (US).

(72) Inventors: CHASSER, Anthony, M.; 1409 Fernledge  
Drive, Allison Park, PA 15101 (US). DUFFY, Shawn, P.;  
325 Linden Drive, Cheswick, PA 15024 (US). AMBROSE,  
Ronald, R.; 6307 Hampton Street, Pittsburgh, PA 15206  
(US).

(74) Agents: MEYERS, Diane, R. et al.; PPG Industries, Inc.,  
One PPG Place, Pittsburgh, PA 15272 (US).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU,  
AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU,  
CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,  
GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,  
LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,  
MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE,  
SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC,  
VN, YU, ZA, ZM, ZW.

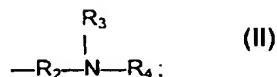
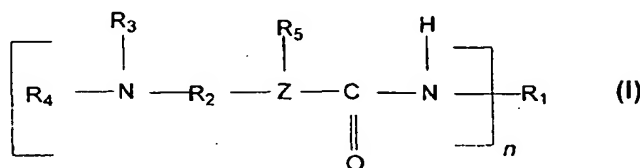
(84) Designated States (*regional*): ARIPO patent (GH, GM,  
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW),  
Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM),  
European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE,  
ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO,  
SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM,  
GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

**Published:**

— without international search report and to be republished  
upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guid-  
ance Notes on Codes and Abbreviations" appearing at the begin-  
ning of each regular issue of the PCT Gazette.

(54) Title: LOW-CURE POWDER COATINGS AND METHODS FOR USING THE SAME



(57) Abstract: Low-cure powder coating compositions are disclosed. The compositions comprise a polyeepoxide and a material having the structure: formula (I) wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms;  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms; Z is oxygen or nitrogen, and when Z is oxygen  $R_5$  is absent and when Z is nitrogen  $R_5$  is hydrogen or is formula (II) and n is 1 to 4. The material can optionally be reacted with an acidic hydrogen-containing compound. The compositions are curable without the use of crosslinking agents or accelerators. Methods for coating a substrate using these compositions, and substrates coated thereby, are also disclosed, as are additional catalysts useful for the same purpose.

WO 03/102044 A2

## **LOW-CURE POWDER COATINGS AND METHODS FOR USING THE SAME**

5

### **FIELD OF THE INVENTION**

The present invention relates to powder coating compositions; more particularly, the present invention relates to low temperature cure thermosetting powder coating compositions. The compositions consistently produce coatings having desirable performance properties when cured, and  
10 that are stable when uncured.

### **BACKGROUND OF THE INVENTION**

Coating compositions have long been used to provide the surface of articles with certain desired physical characteristics, such as color, gloss and  
15 durability. Many coating compositions rely on a liquid carrier, which evaporates after the composition is applied. In recent years, powder coatings have become increasingly popular; because these coatings are inherently low in volatile organic content (VOCs), their use reduces air emissions during the application and curing processes as compared with liquid coatings.

20 Powder coatings are typically cured by heating the coated substrate to an elevated temperature. These temperatures almost always exceed 125°C, and commonly reach about 190°C to 205°C. During the curing process, the powder particles melt and spread, and the components of the powder coating react. In addition to not emitting any VOCs into the environment during the  
25 application or curing processes, powder coating systems are extremely efficient since there is essentially no waste (i.e., application yield is approximately 100 percent). Because of the relatively high (i.e., greater than 125°C) cure temperatures of most powder coatings, their use, for practical purposes, is often limited to substrates that can withstand such high  
30 temperatures or that can be heated to an appropriate temperature long enough for cure to occur.

Despite the desirability of low-cure powder compositions, two problems have prevented their widespread production and use--their mechanical

stability and their chemical stability. Powders that use resins with a glass transition temperature ("Tg") lower than 60°C usually encounter package stability problems, especially if exposed to prolonged heat exposure, and become fused, sintered or clumpy within days. Similarly, prolonged heat exposure can destroy the chemical stability of a powder if it includes crosslinkers that react at temperatures below about 170°C; if a crosslinker with a lower cure temperature is used, cure may be initiated during storage even though the film has not been formed. The premature gelation that occurs in these powder formulations results in coatings having shortened gel times. It is not unusual for low-cure powders to lose >50 percent of their gel time as a result of the premature gelation.

Problems encountered when a powder loses either mechanical or chemical stability can be severe. Poor mechanical stability creates obvious handling, application and appearance issues. Poor chemical stability creates subtler yet just as problematic issues. For example, a powder that has poor chemical stability will fluidize and apply like virgin powder, but because it has advanced in reactivity (i.e. undergone some premature gelation), it demonstrates restricted flow or no flow at all during cure. The result can be a coating having an "orange peel" appearance, a rough texture or gel bodies.

Ideally, a powder should not lose its handling properties under elevated temperature storage and the gel time should remain the same as that of the virgin material. To achieve this, powders are typically formulated with resins having a Tg greater than about 60°C and/or crosslinkers that react at temperatures of about 170°C or greater. Such powders, however, are not low cure. Low-cure powders having lower Tg resins or lower temperature crosslinkers can require expensive storage under refrigeration and air-conditioned application facilities to overcome inherent lack of stability, or must be prepared using special techniques.

Thus, there is a need in the coatings art for low-cure powder coatings having a wide range of application, which also have an acceptable level of

durability when cured on the finished product and good stability at room temperature.

### SUMMARY OF THE INVENTION

5       The present invention is directed to powder coating compositions generally comprising a tertiary aminourea compound, a tertiary aminourethane compound, or mixtures thereof, and a film-forming polyepoxide resin. It has been surprisingly discovered that polyepoxide resins, when used with the present tertiary aminourea and/or aminourethane  
10 compositions, cure to form a suitable coating without the aid of crosslinkers, accelerators, or other additives typically regarded in the art as being necessary to cure a polyepoxide resin. The cured coatings that result from the present compositions have performance properties that are at least as good as powder coating compositions prepared with the same polyepoxides  
15 and conventional curing agents, but lacking the tertiary aminourea or aminourethane compositions described herein. Significantly, this desirable result is achieved by using curing temperatures much lower than those used for conventional products. Accordingly, the present compositions are low-cure. "Low-cure" as used herein refers to powder coating compositions that  
20 cure at a temperature between about 80°C and 125°C. However, the present invention is not limited to this temperature range and also provides cured films at temperatures up to and even greater than 190°C.

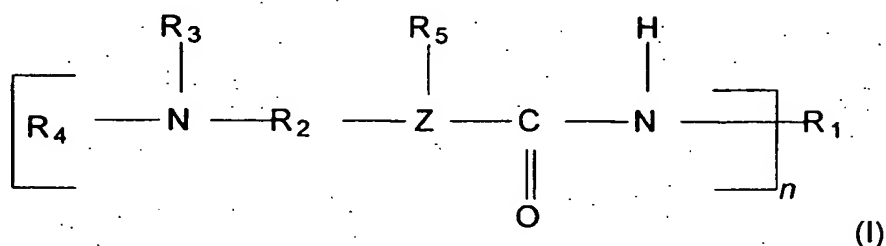
As a result of being low-cure, the present compositions can be used on substrates that are not appropriately exposed to temperatures greater than  
25 about 125°C. Examples include, but are not limited to, plastics such as thermoset and thermoplastic compositions, wood, and pieces of thick metal that cannot be heated above about 95°C because of their size. Also suitable are articles of manufacture that include a variety of substrates; for example, motors that contain both metal and rubber components can be suitably coated  
30 using the present, low-cure powder compositions.

The present compositions also overcome some of the difficulties that have been observed with other powder coating compositions, particularly other low-cure powders. For example, the present powder compositions are storage stable, and reduce, if not eliminate, the problems with chemical and mechanical stability seen with other low-cure powder compositions. The present compositions can be stored at room temperature, and they do not continue to catalyze the reaction of the polyepoxide molecules after the removal of heat. Moreover, the present powder compositions can be prepared using standard methods known in the art for preparing powder coating compositions; no special processing or handling is needed. Thus, the present compositions provide a significant advance in the low-cure powder coatings art.

Methods for coating substrates using the present powder compositions, and substrates coated thereby, are also within the scope of the present invention. Various low-cure catalysts are also included in the present invention.

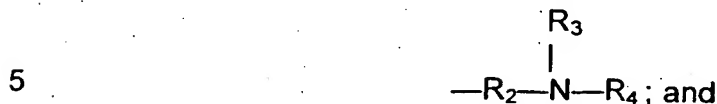
### DETAILED DESCRIPTION OF THE INVENTION

The present invention is directed to a powder coating composition comprising: (a) a material having the structure of Formula I:



25

phenyl groups having 1 to 8 carbon atoms; Z is oxygen or nitrogen and when Z is oxygen R<sub>5</sub> is absent and when Z is nitrogen R<sub>5</sub> is hydrogen or



and *n* is 1 to 4; and (b) a polyepoxide. It will be understood that when Z is oxygen a tertiary aminourethane compound is represented and when Z is nitrogen, Formula I depicts a tertiary aminourea compound. It will be further understood that when R<sub>5</sub> is



there will be two each of R<sub>2</sub>, R<sub>3</sub> and R<sub>4</sub>. Each R<sub>2</sub>, each R<sub>3</sub> and each R<sub>4</sub> can be the same or different as the other R<sub>2</sub>, R<sub>3</sub> or R<sub>4</sub>. For example, one R<sub>2</sub> can have one carbon and the other have two carbons, and the like.

The material of Formula I can be an oligomer wherein R<sub>1</sub> is a monovalent, divalent, trivalent or tetravalent organic radical; divalent radicals are particularly suitable. The R<sub>1</sub> radical can be aliphatic, such as hexamethylene, cycloaliphatic such as cyclohexylene, substituted cycloaliphatic such as 1,1,3,3-tetramethylcyclohexylene, or aromatic such as phenylene. Substituted cycloaliphatics are particularly suitable, especially 1,1,3,3-tetramethylcyclohexylene. Examples of suitable R<sub>2</sub> moieties include ethylene, *n*-propylene, and *iso*- and *n*-butylene. In a particularly suitable composition, Z is nitrogen, R<sub>1</sub> is 1,1,3,3-tetramethylcyclohexylene, R<sub>2</sub> is propylene, R<sub>3</sub> and R<sub>4</sub> are both methyl groups, R<sub>5</sub> is hydrogen and *n* is 2.

The material of component (a) can be prepared by reacting an organic polyisocyanate, particularly diisocyanate, with an amine containing a primary or secondary amine group and a tertiary amine group for the aminourea embodiment or with an alcohol or polyol containing a tertiary amine for the aminourethane embodiment. Suitable polyisocyanates include aliphatic, cycloaliphatic, or aromatic polyisocyanates. Diisocyanates are particularly suitable, although higher polyisocyanates can be used. Examples of suitable

aromatic diisocyanates are 4,4'-diphenylmethane diisocyanate, 1,3-bis(1-isocyanato-1-methylethyl)benzene and derivatives thereof, and toluene diisocyanate. Examples of suitable aliphatic diisocyanates are straight chain aliphatic diisocyanates such as 1,6-hexamethylene diisocyanate  
5 and cycloaliphatic diisocyanates including isophorone diisocyanate and 4,4'-methylene-bis-(cyclohexyl isocyanate). Examples of suitable higher polyisocyanates are 1,2,4-benzene triisocyanate, polymethylene polyphenyl isocyanate and the isocyanurate of isophorone diisocyanate. Isophorone diisocyanate is especially suitable.

10 Examples of amines containing a primary or secondary amine group and a tertiary amine group are dimethylaminopropylamine, bis(dimethylamino)propylamine and 2-amino-5-diethylaminopentane. An example of an alcohol containing a tertiary amine is dimethylaminopropanol. Dimethylaminopropylamine is particularly suitable.

15 The diamine or amino alcohol and polyisocyanate are combined in an equivalent ratio of about 1:1. The diamine is heated to about 50°C, and the polyisocyanate is added over a period of time in the range of about one to two hours, usually about two hours. The amino alcohol typically should be heated to about 80°C before the polyisocyanate is added. The temperature of the  
20 reaction mixture generally increases and is held at an elevated temperature, such as 130°C to 170°C, until the polyisocyanate is completely reacted.

The present invention is further directed to a curable powder composition comprising a polyepoxide and the reaction product of a polyisocyanate and either an amine comprising a primary or secondary amine  
25 group and a tertiary amine, or an alcohol or polyol containing a tertiary amine. Suitable amines and alcohols/polyols, and the method for preparing such a reaction product are described above.

In one embodiment, the material of component (a) further comprises an acidic hydrogen-containing compound; for example, component (a) can  
30 comprise the reaction product of (i) a compound having Formula I and (ii) an acidic hydrogen-containing compound. The acidic hydrogen-containing

compound of (ii) may be a carboxylic acid, a phenolic compound, a polyester, a polyurethane or an acrylic polymer. Phenolic compounds, especially polyphenols, are particularly suitable. Examples of suitable acidic hydrogen-containing compounds include benzoic acid, dodecanedioic acid, azelaic acid, itaconic acid, sebacic acid, and adipic acid. Suitable phenols include phenol itself and polyphenols such as resorcinol, catechol, hydroquinone, bis(4-hydroxyphenyl)-2,2-propane (Bisphenol A), bis(4-hydroxyphenyl)-1,1-isobutane, bis(4-hydroxyphenyl)-1,1-ethane, bis(2-hydroxyphenyl)-methane, 4,4-dihydroxybenzophenone, and 1,5-dihydroxynaphthalene. Bisphenol A is especially suitable.

The reaction product used in the coatings of the present invention can be prepared by mixing the material having Formula 1 of (i) with the acidic hydrogen-containing compound of (ii) in an equivalent ratio of about 1:1 to 1:2, such as about 1:1.87. The material of (i) is typically heated to a temperature of about 140 to 180°C and the acidic hydrogen-containing compound of (ii) is added. The reaction mixture is then usually held at the elevated temperature until it turns clear, indicating homogeneity of the reaction mixture. The reaction mixture is then allowed to cool.

Component (a) in the present compositions, both with and without the acidic hydrogen-containing compound, is used as a catalyst, and typically has a melting point of between about 23°C and 150°C, such as between about 50 and 100°C. This range of melting points helps prevent any curing from taking place in the composition before the application of heat. This improves the long-term stability of curable compositions in which component (a) is used. The melting point of the catalyst is typically not so high, however, that the present compositions lose their characterization as "low-cure". It is therefore desirable that the catalyst used in the present compositions have a melting point of between about 23°C and 150°C; if the melting point were too far above this number, the composition might not cure in the desired manner, and at temperatures too much below this temperature, the composition may not be as stable.



The polyepoxides used in the present compositions are those that are suitable for use in powder coatings, such as those that contain at least two 1,2-epoxide groups per molecule. In general, the epoxy equivalent weight can range from about 180 to about 4000 based on solids of the polyepoxide, 5 such as between about 500 and 1000. The polyepoxides may be saturated or unsaturated, and may be aliphatic, alicyclic, aromatic, or heterocyclic. They may contain substituents such as halogens, hydroxyl groups, and ether groups.

Suitable classes of polyepoxides include epoxy ethers obtained by 10 reacting an epihalohydrin such as epichlorohydrin with a polyphenol in the presence of an alkali. Suitable polyphenols include resorcinol, catechol, hydroquinone, bis(4-hydroxyphenyl)-2,2-propane (Bisphenol A), bis(4-hydroxyphenyl)-1,1-isobutane, bis(4-hydroxyphenyl)-1,1-ethane, bis(2-hydroxyphenyl)-methane, 4,4-dihydroxybenzophenone, and 15 1,5-dihydroxynaphthalene. The diglycidyl ether of Bisphenol A is especially suitable.

Other suitable polyepoxides include polyglycidyl ethers of polyhydric alcohols. These compounds may be derived from polyhydric alcohols such as ethylene glycol, propylene glycol, butylene glycol, 1,6-hexylene glycol, 20 neopentyl glycol, diethylene glycol, glycerol, trimethylol propane, and pentaerythritol. These compounds may also be derived from polymeric polyols such as polypropylene glycol.

Examples of other suitable polyepoxides include polyglycidyl esters of polycarboxylic acids. These compounds may be formed by reacting 25 epichlorohydrin or another epoxy material with an aliphatic or aromatic polycarboxylic acid such as succinic acid, adipic acid, azelaic acid, sebacic acid, maleic acid, 2,6-naphthalene dicarboxylic acid, fumaric acid, phthalic acid, tetrahydrophthalic acid, hexahydrophthalic acid, or trimellitic acid. Dimerized unsaturated fatty acids containing about 36 carbon atoms (Dimer 30 Acid) and polymeric polycarboxylic acids such as carboxyl terminated

acrylonitrile-butadiene rubber may also be used in the formation of these polyglycidyl esters of polycarboxylic acids.

Polyepoxides derived from the epoxidation of an olefinically unsaturated alicyclic compound are also suitable for use in the curable composition of the present invention. These polyepoxides are nonphenolic and are obtained by epoxidation of alicyclic olefins with, for example, oxygen, perbenzoic acid, acid-aldehyde monoperacetate, or peracetic acid. Such polyepoxides include the epoxy alicyclic ethers and esters well known in the art.

Other suitable polyepoxides include epoxy novolac resins. These resins are obtained by reacting an epihalohydrin with the condensation product of aldehyde and monohydric or polyhydric phenols. A typical example is the reaction product of epichlorohydrin with a phenol-formaldehyde condensate.

The curable composition of the present invention may contain one polyepoxide or mixtures of polyepoxides.

Typically, the polyepoxide is present in the curable composition of the present invention in a range of from about 20 to about 90 percent, such as about 30 to 60 percent, based upon total weight of the curable composition.

The catalyst or reaction product is typically present in the compositions of the invention in a range of from about 0.5 to 10 weight percent, such as 3 to 5 weight percent. It is expected that the rate of cure increases as the concentration of catalyst increases, and that these increases are directly proportional. It is surprising, however, that no decrease in chemical or mechanical stability is noted as higher catalyst levels are used; stability often behaves inversely proportional to reactivity, in that as reactivity increases, stability decreases. This maintained stability with increased reactivity is yet another advantage of the present invention.

The powder coating compositions of the present invention may optionally contain additives such as waxes for flow and wetting, flow control agents, such as poly(2-ethylhexyl)acrylate, degassing additives such as

benzoin and MicroWax C, adjuvant resin to modify and optimize coating properties, antioxidants and the like. These optional additives, when used, can be present in amounts up to 10 weight percent, based on total weight of the coating composition, and if used will typically comprise about 1 to 5 weight percent. Any of various pigments standardly used in the powder coatings art can also be included. Pigment weight can be up to 80 percent of the weight of the entire coating and usually is around 35 weight percent of the coating. The compositions can further comprise a plurality of particles, such as organic or inorganic particles, or mixtures thereof, that contribute to the mar and/or scratch resistance of the coatings. Such particles are described in Serial No. 10/007,149, filed on December 5, 2001, which is hereby incorporated by reference. Pigments or solid additives in nanoparticulate form can also be included in the present compositions for the same purpose.

It is both a significant and surprising discovery that the present compositions will cure at low temperatures in the absence of any additional components, such as a crosslinking agent and/or accelerator typically used in conjunction with polyepoxide resins, and thought to be required. In some cases, the use of a crosslinker and accelerator can actually raise the temperature required to cure the polyepoxide, so their use may be undesirable for a low-cure product. Although the inventors do not wish to be bound by any mechanism, it is believed that the reaction product or catalyst used in the present composition catalyzes the reaction of the polyepoxide molecules with themselves. This is in contrast to the standard mechanism of action, in which such a catalyst would be expected to facilitate the reaction between the polyepoxide and crosslinking agent. Thus, the present invention is further directed to a method for initiating self cure of a polyepoxide by adding any of the catalysts described herein to a composition comprising a polyepoxide.

Notwithstanding the lack of a crosslinking agent, the crosslinked density of the present coating compositions can still be controlled to a large extent. This is accomplished by controlling the amount of catalyst added to

the composition. Higher amounts of catalyst usually gel the films faster and crosslink the films more efficiently. In addition, there is a cost savings associated with the elimination of crosslinkers and accelerators, and the ability to cure at a lower temperature. Significantly, the present crosslinker-free and accelerator-free compositions result, upon curing, in coating compositions that have performance properties at least equal to that of conventional powder coatings in which a polyepoxide and conventional crosslinker are used. This refers to the ability to maintain appearance as measured by a number of properties relevant to cured coatings, such as resistance to solvents, pencil hardness, and impact and corrosion resistance.

The present compositions can be prepared by standard methods known in the art. For example, the components are first thoroughly mixed to ensure spatial homogeneity of the ingredients. The composition is then intimately melt kneaded in an extruder. Typical zone temperatures during extrusion range from 40°C to 125°C, such as 45°C to 100°C. The exiting extrudate is rapidly cooled to terminate polymerization. The resulting chip is then micronized into powder with an average particle size of 0.1 to 200 microns, such as 1 to 100 microns. Comminution methods are well known, comminution can be accomplished, for example, by air-classifying mills, impact mills, ball mills, or other fracture-induced mechanisms. Post additives that improve fluidization of the powder mass and/or improve the resistance to impact fusion may be incorporated into the final product before or after micronization. As noted, the use of standard powder coating preparation methods is another advantage of the present invention.

Accordingly, the present invention is further directed to powder coating compositions that cure at a temperature of between 80°C and 125°C comprising a resin and curing agent and wherein substantially all of the curing agent is extruded with the resin. "Substantially all" means the amount of curing agent needed to completely cure the resin. The present invention is further directed to such compositions that do not cure at temperatures below

about 70°C, such as at ambient temperature, like many commercially available low-cure products.

Typically, the present powder coatings will have average particle sizes that range between 15 and 200 microns, such as between about 25 and 50  
5 microns.

The powder coating compositions of the present invention can be applied to a substrate in any number of ways, most often by electrostatic spraying. The powder coating can be applied in a single sweep or in several passes to provide a film having a thickness after cure of from about 1 to 10  
10 mils (25 to 250 microns), usually about 2 to 4 mils (50 to 100 microns). Other standard methods for coating application can also be employed.

After application, the present compositions may be cured by heating to a temperature of between about 80°C and 190°C, preferably between about 80°C and 125°C, for a period ranging from about 3 minutes to 30 minutes,  
15 such as 15 to 20 minutes. Heating can be effected by any means known in the art, typically by placing the coated substrate in an oven. IR radiation can also be used to heat cure the coated substrates.

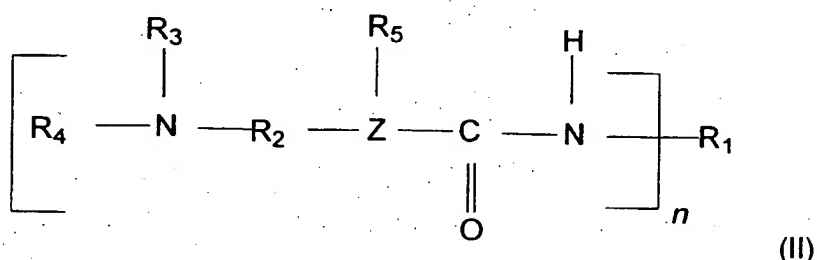
Accordingly, the present invention is further directed to a method for coating a substrate comprising applying to the substrate one or more of the  
20 coating compositions described herein and curing the coating at a temperature of between about 80°C and 190°C, such as between about 80°C and 125°C or between about 105°C and 120°C. In such a method, the polyepoxide will self-cure, or react with itself by homopolymerization; this reaction is catalyzed by the present tertiary aminourea or tertiary  
25 aminourethane compositions. Accordingly, the present invention is further directed to a cured coating layer comprising a polyepoxide and one or more of the catalysts described herein, wherein the polyepoxide is self-cured.

A number of substrates are suitable for coating according to the methods of the present invention, including plastics such as thermosets or  
30 thermoplasts, cardboard, paper, wood, metal, particleboard and medium

density fiberboard or mixtures thereof. Substrates coated according to the present methods are also within the scope of the present invention.

The present invention is further directed to a catalyst composition that is the reaction product of:

- 5 (i) a material having the structure of Formula II and
- (ii) an acidic hydrogen-containing compound:



- 10 For Formula II,  $R_1$  is an organic radical having 6 to 25 carbon atoms,  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen and when  $Z$  is oxygen  $R_5$  is absent and when  $Z$  is nitrogen  $R_5$  is hydrogen or

- 15  $\begin{array}{c} R_3 \\ | \\ -R_2-N-R_4 \end{array}$  and  $n$  is 1 to 4, but when  $Z$  is nitrogen,  $R_2$  is an alkylene having between 1 and 4 carbon atoms and  $R_3$  and  $R_4$  are both alkyl groups having between 1 and 4 carbons.  $R_5$  is not hydrogen.

- 20 The present invention is directed to yet another catalyst composition comprising the compound of Formula I as described above, wherein the composition does not include an acidic hydrogen-containing compound. It has surprisingly discovered that compounds having Formula I function as low-cure catalysts even in the absence of acidic hydrogen-containing compounds.

- 25 As used herein, unless otherwise expressly specified, all numbers such as those expressing values, ranges, amounts or percentages may be read as if prefaced by the word "about", even if the term does not expressly appear. Also, any numerical range recited herein is intended to include all sub-ranges subsumed therein. As used herein, the term "polymer" refers to oligomers

and both homopolymers and copolymers; the prefix "poly" refers to two or more.

### EXAMPLES

- 5        The following examples are intended to illustrate the invention, and should not be construed as limiting the invention in any way.

#### Example 1

- 10        The following ingredients were used to prepare a catalyst of Formula I, wherein an acidic hydrogen-containing compound is used.

| <u>Ingredient</u>                             | <u>Weight, g</u> | <u>Equivalents</u> | <u>Percent by weight</u> |
|---|------------------|--------------------|--------------------------|
| Dimethylaminopropylamine                      | 204.4            | 1.000              | 23.95%                   |
| Isophorone diisocyanate ("IPDI") <sup>1</sup> | 222.2            | 1.000              | 26.05%                   |
| Bisphenol A ("BPA") <sup>2</sup>              | 426.6            | 3.74               | 50.00%                   |

<sup>1</sup>Available from Hüls America, Inc.

<sup>2</sup>4,4'-Isopropylidenediphenol, available from Dow Chemical Co.

- 15        The dimethylaminopropylamine was charged to a suitable reactor and heated to 50°C. The IPDI was added through an addition funnel over a period of two hours. The temperature of the reaction mixture was allowed to increase to 90°C during the addition. After the addition was complete the reaction mixture was heated to 130°C and held at that temperature until infrared analysis indicated consumption of the isocyanate. The reaction mixture was then heated to 160°C and the Bisphenol A was added. The reaction mixture was held at 160°C until the solution turned clear, indicating complete melting of the Bisphenol A. The reaction mixture was poured out hot and allowed to cool and solidify. The final solid product had a solids content of about 98 percent and a number average molecular weight of 336 as measured by gel permeation chromatography using polystyrene as a standard.
- 25

### Example 2

The following ingredients were used to prepare a catalyst of Formula I, wherein an acidic hydrogen-containing compound is not used.

| <u>Ingredient</u>              | <u>Weight, g</u> | <u>Equivalents</u> | <u>Percent by weight</u> |
|--------------------------------|------------------|--------------------|--------------------------|
| Dimethylaminopropylamine       | 204.4            | 1.000              | 47.9%                    |
| Isophorone diisocyanate (IPDI) | 222.2            | 1.000              | 52.1%                    |

- 5        The dimethylaminopropylamine was charged to a suitable reactor and heated to 50°C. The IPDI was added through an addition funnel over a period of two hours. The temperature of the reaction mixture was allowed to increase to 90°C during the addition. After the addition was complete the reaction mixture was heated to 130°C and held at that temperature until
- 10    infrared analysis indicated consumption of the isocyanate. The reaction mixture was poured out hot and allowed to cool and solidify. The final solid product had a solids content of about 98 percent and a number average molecular weight of 336 as measured by gel permeation chromatography using polystyrene as a standard.

15

### Example 3

- Samples 1 to 4 were prepared using the components and amounts shown in TABLE 1, including the products prepared according to Examples 1 and 2. The coatings were prepared by premixing the ingredients in a three-
- 20    blade mixer rotating at 3500 rpm. The premix was then extruded in a 19 mm dual screw extruder operating at a temperature of 80°C. The extrudate was rapidly cooled and pressed into chip. The chip was micronized to an average particle size of 35 microns using a Hosokawa Air-Classifying Mill (ACM).



**TABLE 1**

|                                    | Sample 1 | Sample 2 | Sample 3 | Sample 4 |
|------------------------------------|----------|----------|----------|----------|
| EPON 1001 <sup>3</sup>             |          |          | 340 g    | 340 g    |
| EPON 2002 <sup>4</sup>             |          |          | 140 g    | 140 g    |
| DER 642 <sup>5</sup>               |          | 480 g    |          |          |
| PD 9060 (GMA Acrylic) <sup>6</sup> | 480 g    |          |          |          |
| Product of Example 1               | 15 g     | 15 g     | 15 g     |          |
| Product of Example 2               |          |          |          | 7.5 g    |
| Benzoin <sup>7</sup>               | 4 g      | 4 g      | 4 g      | 4 g      |
| Modaflow <sup>8</sup>              | 9 g      | 9 g      | 9 g      | 9 g      |
| Goresil 210 <sup>9</sup>           | 50 g     | 50 g     | 50 g     | 50 g     |
| TiO <sub>2</sub>                   | 150 g    | 150 g    | 150 g    | 150 g    |

<sup>3</sup> EPON 1001 is a BPA epoxy, standard hybrid type, with an EW=550 from Resolution Performance Products.

5 <sup>4</sup> EPON 2002 is a BPA epoxy, standard hybrid type, with an EW=750 from Resolution Performance Products.

<sup>5</sup> DER 642 is a NOVALAC resin from Dow Chemical.

<sup>6</sup> PD 9060 is a GMA Acrylic resin from Anderson Development.

<sup>7</sup> Added as a degasser.

10 <sup>8</sup> An acrylic copolymer flow additive, anti-crater additive, from Solutia, Inc.

<sup>9</sup> Silica particles, average particle size 2 microns, largest particle size 10 microns, from CED Process Minerals, Inc.

The coatings were sprayed onto Bonderite 1000 steel panels and  
 15 cured at 115.6°C for 25 minutes. Following cure, the panels were subjected to a number of tests standard in the industry for testing coatings. Tests and results are shown in TABLE 2.

**TABLE 2**

|   | Sample 1   | Sample 2   | Sample 3   | Sample 4   |
|---|------------|------------|------------|------------|
| 100 MEK double rubs <sup>10</sup>         | No scuff   | No scuff   | No scuff   | No scuff   |
| Impact Reverse/Direct <sup>11</sup>       | <20/<20    | 70/100     | 160/160    | 160/160    |
| QUV 340 400 hrs <sup>12</sup>             | 60 → 60    | 60 → 20    | 60 → 15    | 60 → 15    |
| Appearance <sup>13</sup>                  | PCI=1      | PCI=7      | PCI=7      | PCI=6      |
| Gel time <sup>14</sup>                    | 6:00       | 3:00       | 3:00       | 3:00       |
| 1000 hrs salt fog 100F <sup>15</sup>      | 2mm creep  | <1mm creep | <1mm creep | <1mm creep |
| 1000 hrs cond hum 100F <sup>16</sup>      | <1mm creep | <1mm creep | <1mm creep | <1mm creep |
| Powder stability (chemical) <sup>17</sup> | 6:00       | 3:00       | 3:00       | 3:00       |

<sup>10</sup> Powder Coatings Institute ("PCI") #8 Recommended Procedure. ("No scuff" means the coating is fully cured.)

<sup>11</sup> ASTM D2794 (Range <20 to 160 in\*lbs.; 160 in\*lbs = full flexibility).

<sup>12</sup> ASTM D4587 results reported in 20° gloss readings taken initially after 400 hours of QUV exposure. No change in 20° gloss = no effect.

<sup>13</sup> PCI visual standards (Range 1 to 10 – 10 being the smoothest).

<sup>14</sup> PCI #6 Recommended Procedure, gel time reported in minutes:seconds.

<sup>15</sup> ASTM B117 (<1 mm = no salt fog effect).

<sup>16</sup> ASTM D1735 (<1 mm creep = no humidity effect).

<sup>17</sup> PCI #1 Recommended Procedure at 32°C; stability reported in minutes:seconds.

The results in TABLE 2 confirm that a variety of polyepoxy resins can be cured at low temperature according to the present invention. The acrylic sample (Sample 1) performed as would be expected in the impact and QUV testing – that is, very poorly on the former and very well on the latter. Bisphenol A epoxies (Samples 3 and 4) work especially well with the present invention providing the highest level of impact resistance, humidity and salt fog resistance, and chemical resistance, but QUV results were low, as would be expected with this type of resin. One skilled in the art could choose the appropriate resin based on the desired qualities of the cured coating, using the present catalysts to effect cure at low temperatures.

#### Example 4

Sample 3 prepared as described above was tested for stability using standard techniques as discussed below. The stability of Sample 3 was also

compared with the stability of Sample 5, prepared in the same manner as Sample 3 except using three grams of 2-methyl imidazole as the catalyst instead of the catalyst prepared according to Example 1. A standard polyepoxide resin cured with an acid polyester was also compared (PCF

- 5 80147, commercially available from PPG Industries, Inc.). The coatings were applied as described in Example 2. However, the commercially available product was cured at a higher temperature (162.8°C) compared to 115.6°C for Sample 3 and Sample 5.

10

**TABLE 3**

|   | PCF 80147 | Sample 3  | Sample 5  |
|---|-----------|-----------|-----------|
| Mechanical Stability<br>One week at 32°C  | Excellent | Excellent | Excellent |
| Chemical stability<br>Initial Gel @ 145°C | 4:00      | 3:00      | 1:30      |
| Gel after One Week @<br>32°C              | 4:00      | 3:00      | :40       |
| 100 MEK Double rubs                       | No Scuff  | No Scuff  | No Scuff  |

- The chemical stability and mechanical stability tests were identical, and were performed by placing virgin, free-flowing powder in a sealed jar and setting the jar in a water bath heated (PCI #1 Recommended Procedure, as
- 15 described in TABLE 2). After one week the samples were evaluated for mechanical stability using a visual ranking. A free-flowing powder is excellent; the ranking standardly used in the industry is as follows:  
excellent>good>cakey>clumpy>fused>sintered. All samples had an excellent mechanical stability.

20

After the visual ranking for mechanical stability, gel times of the aged powder were taken as per PCI #6 Recommended Procedure to assess the chemical stability of the powder coating. A slower gel time translates to advancement in molecular weight. A powder coating should not have molecular weight advancement during storage. As shown in TABLE 3, only

Sample 5 (2-methyl imidazole catalyst) showed advancement; the commercial product and the product of the present invention did not advance over time.

Solvent cure (100 MEK double rubs – PCI #8 Recommended Procedure) was used as an indication of film cure. When a film has excellent  
5 solvent resistance, that is a good indication that complete cure has occurred. Sample 3 of the present invention underwent complete cure just as the other samples tested.

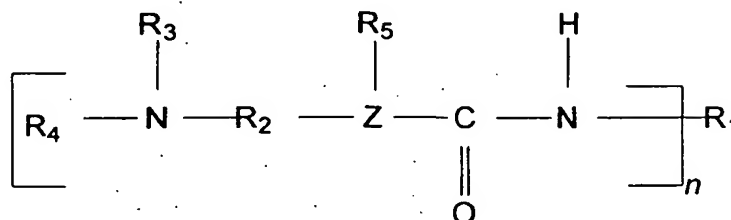
Thus, the low-cure composition of the present invention performed equal to a commercially available high cure product using conventional  
10 crosslinkers and performed better than a sample using a low temperature curing agent outside the scope of the present invention.

Whereas particular embodiments of this invention have been described above for purposes of illustration, it will be evident to those skilled in the art  
15 the numerous variations of the details of the present invention may be made without departing from the invention as defined in the appended claims.

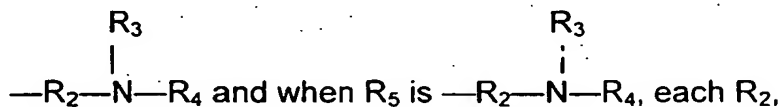
THEREFORE, WHAT IS CLAIMED IS:

1. A curable powder composition comprising:

(a) a material having the structure:



wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms,  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen and when  $Z$  is oxygen,  $R_5$  is absent and when  $Z$  is nitrogen,  $R_5$  is hydrogen or



each  $R_3$  and each  $R_4$  are the same or different; and  $n$  is 1 to 4; and

(b) a polyepoxide;

wherein the powder composition cures in the absence of any addition component.

2. The powder composition of Claim 1, wherein said composition cures at a temperature of between 80°C and 125°C.

3. The powder composition of Claim 2, wherein said composition cures at a temperature of between 105°C and 120°C.

4. The powder composition of Claim 1, wherein said composition cures at a temperature greater than 125°C.

5. The powder composition of Claim 1, wherein Z is nitrogen and R<sub>5</sub> is hydrogen.
- 5 6. The composition of Claim 5, wherein R<sub>1</sub> is 1,1,3,3-tetramethylcyclohexylene.
7. The composition of Claim 5, wherein R<sub>2</sub> is n-propylene.
- 10 8. The composition of Claim 5, wherein R<sub>3</sub> and R<sub>4</sub> are methyl.
9. The composition of Claim 1, wherein component (a) further comprises an acidic hydrogen-containing compound.
- 15 10. The composition of Claim 9, wherein the acidic hydrogen-containing compound is a phenolic compound.
11. The composition of Claim 10, wherein the phenolic compound is a polyphenol.
- 20 12. The composition of Claim 11, wherein the polyphenol is bis(4-hydroxyphenyl)-2,2-propane.
13. The composition of Claim 1, wherein (a) is present in an amount  
25 ranging from about 0.5 to 10 weight percent, and (b) is present in an amount ranging from about 20 to about 90 weight percent, with weight percent being based upon total weight of the composition.
14. The composition of Claim 12, wherein (a) is present in an  
30 amount ranging from about 3 to 5 weight percent, and (b) is present in an

amount ranging from about 30 to about 60 weight percent, with weight percent being based upon total weight of the composition.

15. A method for coating a substrate comprising:
- 5 (a) applying to said substrate the powder composition of Claim 1; and
- (b) curing said composition.

16. The method of Claim 15, wherein said composition is cured at a
- 10 temperature of between 80°C and 125°C.

17. A substrate coated according to the method of Claim 15.

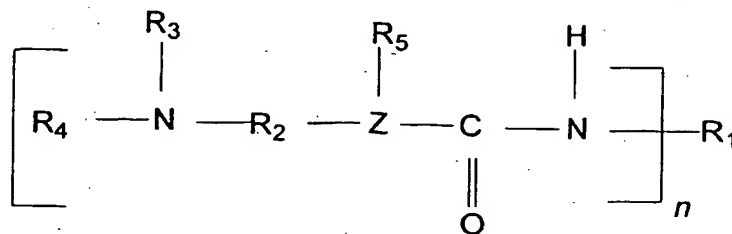
18. The substrate of Claim 17, wherein said substrate is comprised
- 15 of a thermoplastic polymer, a thermoset polymer, cardboard, paper, wood, particleboard, medium density fiberboard and/or metal.

19. A powder coating composition that cures at a temperature of between 80°C and 125°C, comprising a resin and curing agent therefor,
- 20 wherein substantially all of the curing agent is extruded with the resin.

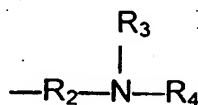
20. The powder coating composition of Claim 19, wherein said composition does not cure at a temperature below 70°C.

- 25 21. A catalyst composition comprising:
- the reaction product of:

i) a material having the structure



wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen and when  $Z$  is oxygen  $R_5$  is absent and when  $Z$  is nitrogen,  $R_5$  is hydrogen or

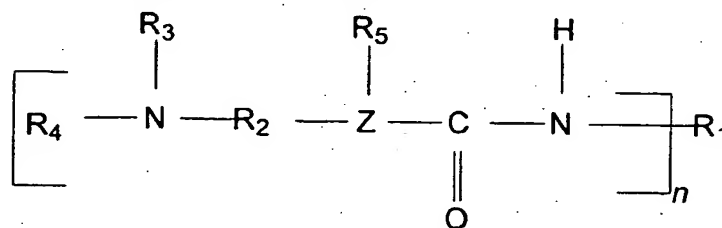


and when  $R_5$  is  $\begin{array}{c} R_3 \\ | \\ -R_2 - N - R_4 \end{array}$ , each  $R_2$ , each  $R_3$  and each  $R_4$  are the same or different, and  $n$  is 1 to 4, but when  $Z$  is nitrogen,  $R_2$  is an alkylene having between 1 and 4 carbon atoms and  $R_3$  and  $R_4$  are both alkyl groups having between 1 and 4 carbons.  $R_5$  is not hydrogen; and

(ii) an acidic hydrogen-containing compound;

22. A powder composition comprising the catalyst of Claim 21.

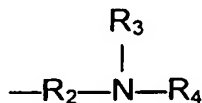
23. A catalyst composition comprising:  
a material having the structure:





wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms,  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen and when  $Z$  is oxygen  $R_5$  is absent and when  $Z$  is nitrogen  $R_5$  is hydrogen or

5



10

and when  $R_5$  is  $-R_2-N(R_3)-R_4$ , each  $R_2$ , each  $R_3$  and each  $R_4$  are the same or different; and  $n$  is 1 to 4; wherein said composition does not include an acidic hydrogen-containing compound.

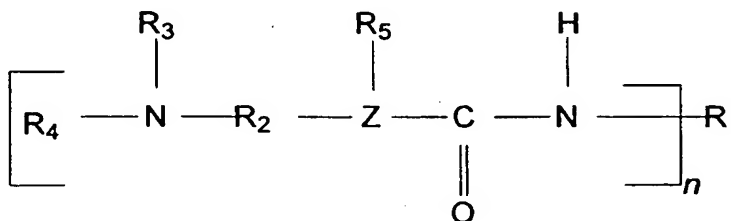
24. A powder composition comprising the catalyst of Claim 23, wherein said powder composition does not include an acidic hydrogen-containing compound.

15

25. A cured coating layer comprising:

(a) a material having the structure:

20



25

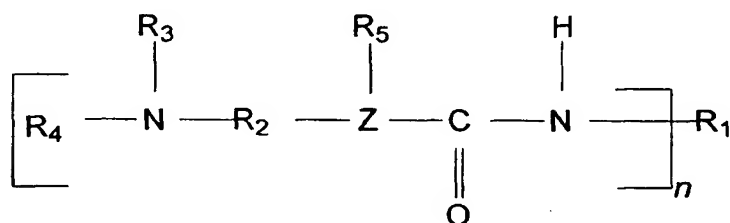
wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms,  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen and when  $Z$  is oxygen,  $R_5$  is absent and when  $Z$  is nitrogen,  $R_5$  is hydrogen or

$$\text{—R}_2\text{—}\overset{\text{R}_3}{\underset{|}{\text{N}}}\text{—R}_4$$
 and when R<sub>5</sub> is  $\text{—R}_2\text{—}\overset{\text{R}_3}{\underset{|}{\text{N}}}\text{—R}_4$ , each R<sub>2</sub>, each R<sub>3</sub> and each R<sub>4</sub> are the same or different; and *n* is 1 to 4; and

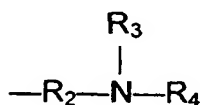
(b) a polyepoxide;

wherein the polyepoxide has reacted with itself during cure.

26. A method for initiating self cure of a polyepoxide resin  
10 comprising adding to a composition comprising said resin a catalyst having  
the structure



wherein R<sub>1</sub> is an organic radical having 6 to 25 carbon atoms, R<sub>2</sub> is an organic radical having 1 to 20 carbon atoms; R<sub>3</sub> and R<sub>4</sub> are independently alkyl or phenyl groups having 1 to 8 carbon atoms; Z is oxygen or nitrogen and when Z is oxygen R<sub>5</sub> is absent and when Z is nitrogen, R<sub>5</sub> is hydrogen or



20 and when R<sub>5</sub> is  $\text{—R}_2\text{—N}(\text{R}_3)\text{—R}_4$ , each R<sub>2</sub>, each R<sub>3</sub> and each R<sub>4</sub> are the same or different; and n is 1 to 4.

25            27. The method of Claim 26, wherein said catalyst further comprises  
an acidic hydrogen-containing compound.

28. The composition of Claim 1, wherein R<sub>2</sub> is alkylene.

29. The composition of Claim 1, wherein  $R_3$  and  $R_4$  are alkyl.

30. A curable powder composition comprising:

- 5 (a) a polyepoxide; and  
(b) the reaction product of a polyisocyanate and either an amine comprising a primary or secondary amine group and a tertiary amine or an alcohol or polyol containing a tertiary amine.

10 31. The composition of Claim 30, wherein the reaction product is formed from a polyisocyanate and an amine comprising a primary or secondary amine group and a tertiary amine.

15 32. The composition of Claim 31, wherein the polyisocyanate is a diisocyanate.

33. The composition of Claim 30, wherein (b) is mixed with an acidic hydrogen-containing compound.

20

***This Page Blank (uspto)***

(19) World Intellectual Property  
Organization  
International Bureau



(43) International Publication Date  
11 December 2003 (11.12.2003)

PCT

(10) International Publication Number  
WO 2003/102044 A3

(51) International Patent Classification<sup>7</sup>: C08G 59/68,  
C07C 275/14, C08L 63/00, C08G 18/28, C09D 5/03

(21) International Application Number:  
PCT/US2003/014366

(22) International Filing Date: 6 May 2003 (06.05.2003)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
10/160,466 31 May 2002 (31.05.2002) US

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(71) Applicant: PPG INDUSTRIES OHIO INC. [US/US];  
3800 West 143rd Street, Cleveland, OH 44111 (US).

(72) Inventors: CHASSER, Anthony, M.; 1409 Fernledge  
Drive, Allison Park, PA 15101 (US). DUFFY, Shawn, P.;  
325 Linden Drive, Cheswick, PA 15024 (US). AMBROSE,  
Ronald, R.; 6307 Hampton Street, Pittsburgh, PA 15206  
(US).

(74) Agents: MEYERS, Diane, R. et al.; PPG Industries, Inc.,  
One PPG Place, Pittsburgh, PA 15272 (US).

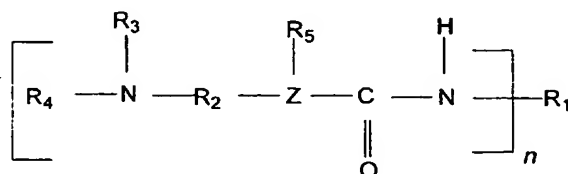
**Published:**

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

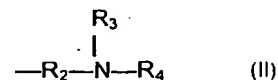
(88) Date of publication of the international search report:  
25 March 2004

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: LOW-CURE POWDER COATINGS AND METHODS FOR USING THE SAME



(I)



(57) Abstract: Low-cure powder coating compositions are disclosed. The compositions comprise a polyeepoxide and a material having the structure: formula (I) wherein  $R_1$  is an organic radical having 6 to 25 carbon atoms;  $R_2$  is an organic radical having 1 to 20 carbon atoms;  $R_3$  and  $R_4$  are independently alkyl or phenyl groups having 1 to 8 carbon atoms;  $Z$  is oxygen or nitrogen, and when  $Z$  is oxygen  $R_3$  is absent and when  $Z$  is nitrogen  $R_3$  is hydrogen or is formula (II) and  $n$  is 1 to 4. The material can optionally be reacted with an acidic hydrogen-containing compound. The compositions are curable without the use of crosslinking agents or accelerators. Methods for coating a substrate using these compositions, and substrates coated thereby, are also disclosed, as are additional catalysts useful for the same purpose.

WO 2003/102044 A3

# INTERNATIONAL SEARCH REPORT

International Application No

T/US 03/14366

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08G59/68 C07C275/14 C08L63/00 C08G18/28 C09D5/03

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07C C08G C08L C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

| Category * | Citation of document, with indication, where appropriate, of the relevant passages   | Relevant to claim No. |
|------------|--|-----------------------|
| X          | US 5 569 733 A (DONNELLY KAREN D ET AL)<br>29 October 1996 (1996-10-29)<br>claim 1; examples A,B,D,F,1,2,4,6; tables<br>I,II | 21,23,<br>26,27       |
| X          | DE 195 12 479 A (BAYER AG)<br>10 October 1996 (1996-10-10)<br>table page 5<br>page 4, line 1 - line 43                       | 23                    |
| X          | EP 0 594 133 A (AJINOMOTO KK)<br>27 April 1994 (1994-04-27)<br>page 5, line 1 -page 7, line 10; examples<br>8-13,15          | 23,26                 |
|            | ---  |                       |
|            | -/--   |                       |

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

"I" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"Z" document member of the same patent family

Date of the actual completion of the international search

13 August 2003

Date of mailing of the international search report

29.01.04

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Meiners, C

# INTERNATIONAL SEARCH REPORT

International Application No

/US 03/14366

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

| Category * | Citation of document, with indication, where appropriate, of the relevant passages   | Relevant to claim No. |
|------------|--|-----------------------|
| X          | US 4 324 739 A (ZONDLER HELMUT ET AL)<br>13 April 1982 (1982-04-13)<br>*abstract*<br>column 4, line 48 -column 16, line 17;<br>tables I,IV | 21,23,<br>25-27       |
| Y          | *whole document*   | 1-18,<br>21-33        |
| Y          | EP 0 138 769 A (CIBA GEIGY AG)<br>24 April 1985 (1985-04-24)<br>*whole document*   | 1-18,<br>21-33        |

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US 03/14366

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☐ Claims Nos.:  
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

see additional sheet

1. ☐ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☒ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

1-18, 21-29, 30-33

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.



FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

This International Searching Authority found multiple (groups of) inventions in this international application, as follows:

1. Claims: 1-18, 21-29, and 30-33

Claims 1-18 and 21-29 are concerned with catalysts compositions comprising urethane and urea derivatives and curable powder (coating) compositions comprising the catalysts according to claims 21 and 23 and formula I of the present application. Furthermore, methods for using the curable compositions and the cured coating layers are claimed.

Claims 30-33 are concerned with curable powder compositions comprising a polyepoxide and the reaction product of a polyisocyanate and either an amine comprising a primary or secondary amine group and a tertiary amine or an alcohol or polyol containing a tertiary amine.

2. Claim : 19 and 20

Claims 19 and 20 are concerned with powder coating compositions that cure at a temperature of between 80 and 125 C. The Compositions of claim 19 comprise a resin and a curing agent therefore, wherein substantially all of the curing agent is extruded with the resin.

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

US 03/14366

| Patent document<br>cited in search report |   | Publication<br>date | Patent family<br>member(s) | Publication<br>date |
|---|---|---------------------|----------------------------|---------------------|
| US 5569733                                | A | 29-10-1996          | WO 9526951 A1              | 12-10-1995          |
| DE 19512479                               | A | 10-10-1996          | DE 19512479 A1             | 10-10-1996          |
| EP 0594133                                | A | 27-04-1994          | JP 3367531 B2              | 14-01-2003          |
|   |   |                     | JP 6211969 A               | 02-08-1994          |
|   |   |                     | JP 3367532 B2              | 14-01-2003          |
|   |   |                     | JP 6211970 A               | 02-08-1994          |
|   |   |                     | DE 69318369 D1             | 10-06-1998          |
|   |   |                     | DE 69318369 T2             | 28-01-1999          |
|   |   |                     | EP 0594133 A2              | 27-04-1994          |
|   |   |                     | US 5430112 A               | 04-07-1995          |
| US 4324739                                | A | 13-04-1982          | AT 6520 T                  | 15-03-1984          |
|   |   |                     | CA 1151345 A1              | 02-08-1983          |
|   |   |                     | DE 3066796 D1              | 12-04-1984          |
|   |   |                     | EP 0017623 A2              | 15-10-1980          |
|   |   |                     | ES 8104353 A1              | 01-07-1981          |
|   |   |                     | FI 801105 A ,B,            | 11-10-1980          |
|   |   |                     | JP 1480011 C               | 10-02-1989          |
|   |   |                     | JP 55153752 A              | 29-11-1980          |
|   |   |                     | JP 63026743 B              | 31-05-1988          |
|   |   |                     | NO 801020 A ,B,            | 13-10-1980          |
|   |   |                     | US 4352913 A               | 05-10-1982          |
| EP 0138769                                | A | 24-04-1985          | CA 1229600 A1              | 24-11-1987          |
|   |   |                     | DE 3471652 D1              | 07-07-1988          |
|   |   |                     | EP 0138769 A2              | 24-04-1985          |
|   |   |                     | JP 60104055 A              | 08-06-1985          |